

BA-22124



CERTIFICATION

I, Noriaki NAGAHAMA, c/o ITOH & CO. located at Toranomoh  
Denki Bldg., 8-1, 2-chome, Toranomoh, Minato-ku, Tokyo, Japan,  
hereby certify that I am the translator of the accompanying  
certified official copy of the documents in respect of an  
application for a patent filed in Japan on December 28, 1989 and  
of the official certificate attached thereto, and certify that  
the following is a true and correct translation to the best of my  
knowledge and belief.

*Noriaki Nagahama*  
Noriaki NAGAHAMA

Dated this 30<sup>th</sup> day of November, 1994

*Noriaki Nagahama*

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This is to certify that the annexed is a true copy  
of the following application as filed with this Office.

Date of Application : December 28, 1989

Application Number : (Hei.) 1-341244

Applicant(s) : NIPPON OIL CO., LTD.

January 16, 1991

Commissioner,

Patent Office Satoshi UEMATSU

Certificate No. (Hei.) 2-76213

C10M 105/38

C10N 40:00

PETITION FOR PATENT 1

December 28, 1989

To Commissioner of the Patent Office

Fumitake YOSHIDA

1. Title of the Invention:

REFRIGERATOR OILS FOR USE WITH CHLORINE-  
FREE TYPE HALOGENOCARBON REFRIGERANTS

2. Number of claims ..... 6

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6. List of Attached Documents:

- |                               |        |
|-------------------------------|--------|
| (1) Specification             | 1 copy |
| (2) Duplicate of the Petition | 1 copy |
| (3) Power of Attorney         | 1 copy |
- (This will be filed later.)

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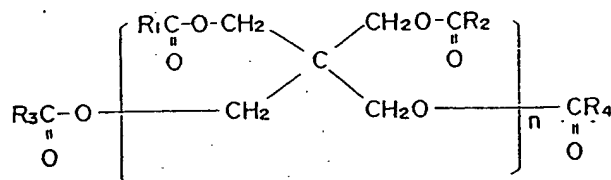
# SPECIFICATION

## 1. Title of the Invention:

REFRIGERATOR OILS FOR USE WITH CHLORINE-FREE TYPE  
HALOGENOCARBON REFRIGERANTS

## 2. What is claimed is:

1. A refrigerator oil for use in compressors using therein a chlorine-free type halogenocarbon as a refrigerant, comprising as a major component a pentaerythritol ester represented by the general formula



wherein  $\text{R}_1\text{-R}_4$  may be identical with, or different from, each other and are each a group selected from the group consisting of straight-chain alkyl groups having 3-11 carbon atoms, branched-chain alkyl groups having 3-15 carbon atoms and cycloalkyl groups having 6-12 carbon atoms, the straight-chain alkyl groups being present in a ratio of not

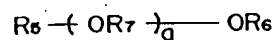
more than 60 % of the total alkyl groups, and n is an integer of 1-3.

2. A refrigerator oil according to claim 1, wherein the pentaerythritol ester is comprised as a base oil.

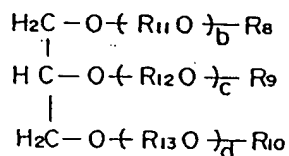
3. A refrigerator oil according to claim 1, comprising as a base oil a mixture oil of:

(I) the pentaerythritol ester, and

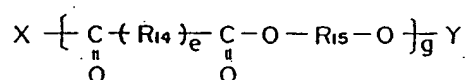
(II) at least one kind of an oil selected from the group consisting of a polyoxyalkylene glycol or an ether thereof represented by the general formula



wherein  $R_5$  and  $R_6$  are each a hydrogen atom or an alkyl group having 1-18 carbon atoms,  $R_7$  is an alkylene group having 2-4 carbon atoms, and a is an integer of 5-70, a polyoxyalkylene glycol glycerol ether represented by the general formula



wherein  $R_8$ - $R_{10}$  are each a hydrogen atom or an alkyl group having 1-18 carbon atoms,  $R_{11}$ - $R_{13}$  are each an alkylene group having 2-4 carbon atoms, and b-d are each an integer of 5-7, an ester represented by the general formula



wherein X is a group represented by the general formula

$$-OR_{16} \text{ or } -O-R_{17}-O-\underset{\underset{O}{|}}{C}-R_{18} \quad , Y \text{ is a group represented by}$$

the general formula  $\begin{array}{c} \text{O} \\ \parallel \\ -\text{C}-\text{R}_{19} \\ \text{O} \end{array}$  or  $\begin{array}{c} \text{O} \\ \parallel \\ -\text{C}-(\text{R}_{20})_f-\text{C}-\text{OR}_{21} \\ \text{O} \end{array}$ ,

$R_{14}$  and  $R_{20}$  are each a divalent hydrocarbon group having 1-8 carbon atoms,  $R_{15}$  and  $R_{17}$  are each a divalent saturated

hydrocarbon group having 2-16 carbon atoms, R<sub>16</sub> and R<sub>21</sub> are

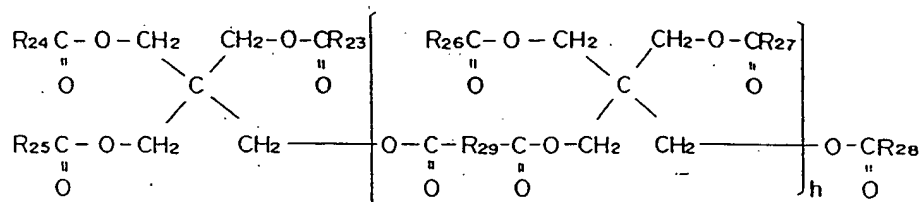
each an alkyl group having 1-15 carbon atoms,  $R_{18}$  and  $R_{19}$

are each an alkyl group having 1-14 carbon atoms, e and f

are each an integer of 0 or 1 and n is an integer of 0-30,

and

a pentaerythritol dicarboxylic acid ester represented by the general formula



wherein  $R_{22}$ - $R_{28}$  are each an alkyl group having 3-15 carbon atoms,  $R_{29}$  is a divalent hydrocarbon group having 1-8 carbon atoms and h is an integer of 1-5.

4. A refrigerator oil according to claim 3, wherein the pentaerythritol ester (I) is comprised in an amount of more than 50 % by weight based on the total amount of the refrigerator oil.

5. A refrigerator oil according to any one of claims 1-4, further comprising at least one kind of a phosphorous compound selected from the group consisting of phosphoric esters, acid phosphoric esters, amine salts of acid phosphoric esters, chlorinated phosphoric esters and phosphorous esters, in an amount of 0.1-5.0 % by weight based on the total amount of the refrigerator oil.

6. A refrigerator oil according to any one of claims 1-5, further comprising at least one kind of an epoxy compound selected from the group consisting of phenylglycidyl ether type epoxy compounds, glycidyl ester type epoxy compounds, epoxidized fatty acid monoesters and epoxidized vegetable oils, in an amount of 0.1-5.0 % by weight based on the total amount of the refrigerator oil.

3. Detailed description of the Invention:



[Field of the Invention]

This invention relates to a lubricating oil for compressors of refrigerators using therein a chlorine-free type halogenocarbon as a refrigerant (the oil being hereinafter referred to as "a refrigerator oil for use with a chlorine-free type halogenocarbon refrigerant") and, more specifically, it relates to such a refrigerator oil which comprises a pentaerythritol ester having a specific chemical structure as a major component and is superior in various properties.

[Prior Art and Problems to be solved by the Invention]

Generally, naphthenic mineral oils, paraffinic mineral oils, alkylbenzenes, polyglycolic oils, ester oils and mixtures thereof, which have each a kinematic viscosity of 10 - 200 cSt at 40°C, as well as said oils incorporated with suitable additives have been used as refrigerator oils.

On the other hand, chlorofluorocarbons (CFCs) type refrigerants, such as CFC-11, CFC-12, CFC-113 and HCFC-22, have been used for refrigerators.

Of these CFCs, CFCs such as CFC-11, CFC-12 and CFC-113, which are obtained by substituting all the hydrogen atoms of hydrocarbons thereof by halogen atoms including chlorine atoms, may lead to the destruction of the ozone layer, and therefore, the use of the CFCs has been controlled. Accordingly, chlorine-free type halogenocarbons, such as HFC-134a and HFC-152a, have been

being used as substitutes for CFCs. HFC-134a is especially hopeful as a substitute refrigerant since it is similar in thermodynamic properties to CFC-12 which has heretofore been used in many kinds of refrigerators of home cold-storage chests, air-conditioners and the like.

Refrigerator oils require various properties, among which their compatibility with refrigerants is extremely important in regard to lubricity and system efficiency in refrigerators. However, conventional refrigerator oils comprising, as the base oils, naphthenic oils, paraffinic oils, alkylbenzenes, heretofore known ester oils and the like, are hardly compatible with chlorine-free type halogenocarbons such as HFC-134a. Therefore, if said conventional refrigerator oils are used in combination with HFC-134a, the resulting mixture will separate into two layers at normal temperature thereby to degrade the oil-returnability which is the most important within the refrigeration system and cause various troubles such as a decrease in refrigeration efficiency, the deterioration of lubricity and the consequent seizure of the compressor within the system whereby the refrigerator oils are made unsuitable for use as such. In addition, polyglycolic oils are also known as refrigerator oils for their high viscosity index and are disclosed in, for example, Japanese Pat. Gazettes Nos. Sho. 57-42119 and Sho. 61-52880 and Japanese Pat. Appln. Laid-Open Gazette No. Sho. 57-51795. However, the polyglycolic oils concretely disclosed in these prior

art publications are not fully compatible with HFC-134a thereby raising the same problems as above and rendering them unusable.

Further, U.S. Patent No. 4,755,316 discloses polyglycolic refrigerator oils which are compatible with HFC-134a. In addition, the present inventors developed polyglycolic refrigerator oils which have excellent compatibility with HFC-134a as compared with conventional known refrigerator oils, and filed applications for patents for the thus developed polyglycolic refrigerator oils (Japanese Pat. Appln. Laid-Open Gazettes Nos. hei. 1-256594 and hei. 1-271491). It has been found, however, that the polyglycolic oils raise problems as to their high compatibility with water and inferior electrical insulating property.

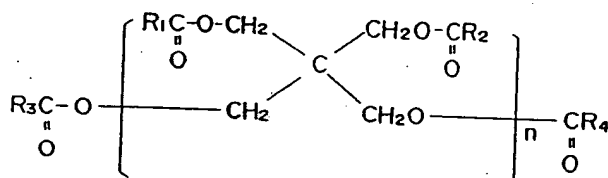
On the other hand, refrigerator oils used in compressors of home refrigerators and the like are required to have a high electrical insulating property. Among the known refrigerator oils, alkylbenzenes and the mineral oils have the highest insulating property, but they are hardly compatible with chlorine-free type halogenocarbons such as HFC-134a as mentioned above. Therefore, there had been developed no refrigerator oils having both high compatibility with chlorine-free type halogenocarbons such as HFC-134a and a high insulating property before the accomplishment of the present invention.

The present inventors made various intensive studies in attempts to develop refrigerator oils which can meet the aforesaid requirements and, as the result of their studies, they found that esters having specific structures have excellent compatibility with chlorine-free type halogenocarbons such as HFC-134a, and a high electrical insulating property as well as excellent lubricity. This invention is based on this finding.

The object of this invention is to provide lubricating oils for use with chlorine-free type halogenocarbons refrigerants, the oils comprising as a major component (or a base oil) an ester having a specific structure and having excellent compatibility with chlorine-free type halogenocarbons such as HFC-134a, and a high electrical insulating property.

[Means of Solution to the Problems]

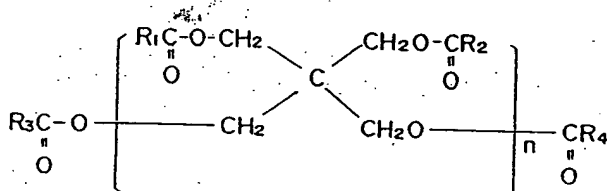
The refrigerator oil of the present invention is characterized in that it comprises as a major component a pentaerythritol ester represented by the general formula



wherein  $R_1-R_4$  may be identical with, or different from, each other and are each a group selected from the group consisting of straight-chain alkyl groups having 3-11 carbon atoms, branched-chain alkyl groups having 3-15 carbon atoms and cycloalkyl groups having 6-12 carbon atoms, the straight-chain alkyl groups being present in a ratio of not more than 60 % of the total alkyl groups, and n is an integer of 1-3.

This invention will be explained in more detail hereunder.

The refrigerator oils of the present invention are those comprising as a major component a pentaerythritol ester represented by the general formula



In the formula,  $R_1$ - $R_4$  may be identical with, or different from, each other and are each a group selected from the group consisting of straight-chain alkyl groups having 3-11 carbon atoms, preferably 3-7 carbon atoms, branched-chain alkyl groups having 3-15 carbon atoms, preferably 4-11 carbon atoms and cycloalkyl groups having 6-12 carbon atoms, preferably 6-8 carbon atoms. The cycloalkyl groups in this invention may include alkylcycloalkyl groups. In addition,

a ratio of the straight-chain alkyl groups to the total alkyl groups is not more than 60%, preferably not more than 50%. Further,  $n$  is an integer of 1-3. Thus, the formula illustrates monopentaerythritol esters, dipentaerythritol esters and tripentaerythritol esters. A refrigerator oil comprising as a major component a pentaerythritol ester which does not satisfy the above conditions is undesirably inferior in miscibility with a chlorine-free type halogenocarbon.

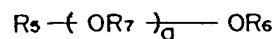
$R_1-R_4$  are each exemplified by  $n$ -propyl group,  $n$ -butyl group,  $n$ -pentyl group,  $n$ -hexyl group,  $n$ -heptyl group,  $n$ -octyl group,  $n$ -nonyl group,  $n$ -decyl group,  $n$ -undecyl group, iso-propyl group, iso-butyl group, iso-pentyl group, iso-hexyl group, iso-heptyl group, iso-octyl group, iso-nonyl group, iso-decyl group, iso-undecyl group, iso-dodecyl group, iso-tridecyl group, iso-tetradecyl group, iso-pentadecyl group, cyclohexyl group, cycloheptyl group, cyclooctyl group, cyclononyl group, cyclodecyl group, cycloundecyl group, cyclododecyl group, methylcyclohexyl group, ethylcyclohexyl group, propylcyclohexyl group, butylcyclohexyl group, pentylcyclohexyl group or hexylcyclohexyl group.

The pentaerythritol esters used in the present invention are esters of pentaerythritol, dipentaerythritol or tripentaerythritol and a monocarboxylic acid, and are ordinarily obtained by the reaction of pentaerythritol, dipentaerythritol, tripentaerythritol or a

mixture thereof with a mixture of at least one carboxylic acid having the aforementioned alkyl group. The products obtained by the methods as mentioned above may be refined to remove the by-products and/or unreacted reactants, but the by-products and/or unreacted reactants may be present in small amounts in the refrigerator oils of the present invention as far as they do not impair the excellent performances thereof. The compounds having the chemical structure represented by the above formula may be used in the present invention as the pentaerythritol esters. The kinematic viscosities of the pentaerythritol esters according to the present invention are in the range of preferably 2-150 cSt, more preferably 5-100 cSt at 100°C.

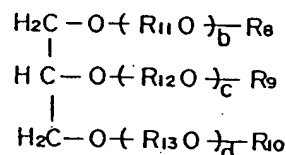
The refrigerator oil of the present invention may comprise as the only base oil the pentaerythritol ester mentioned above and, as required, it may additionally comprise other base oils for refrigerator oils. Among the other base oils, preferable ones are illustrated as follows:

a polyoxyalkylene glycol or an ether thereof represented by the general formula

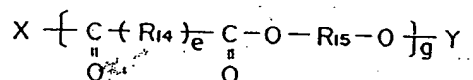


wherein  $R_5$  and  $R_6$  are each a hydrogen atom or an alkyl group having 1-18 carbon atoms,  $R_7$  is an alkylene group having 2-4 carbon atoms and  $a$  is an integer of 5-70.

a polyoxyalkylene glycol glycerol ether represented by the general formula



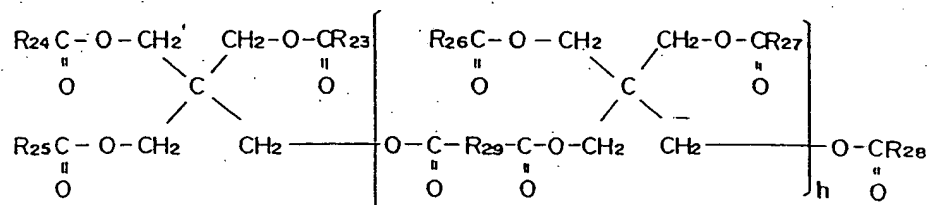
wherein  $\text{R}_8-\text{R}_{10}$  are each a hydrogen atom or an alkyl group having 1-18 carbon atoms,  $\text{R}_{11}-\text{R}_{13}$  are each an alkylene group having 2-4 carbon atoms and b-d are each an integer of 5-7, an ester represented by the general formula



wherein X is a group represented by the general formula  $-\text{OR}_{16}$  or  $-\text{O}-\text{R}_{17}-\text{O}-\underset{\text{O}}{\text{C}}-\text{R}_{18}$ , Y is a group represented by the general formula  $-\underset{\text{O}}{\text{C}}-\text{R}_{19}$  or  $-\underset{\text{O}}{\text{C}}-\left(\text{R}_{20}\right)_f-\underset{\text{O}}{\text{C}}-\text{OR}_{21}$ ,  $\text{R}_{14}$  and  $\text{R}_{20}$  are each a divalent hydrocarbon group having 1-8 carbon atoms,  $\text{R}_{15}$  and  $\text{R}_{17}$  are each a divalent saturated hydrocarbon group having 2-16 carbon atoms,  $\text{R}_{16}$  and  $\text{R}_{21}$  are each an alkyl group having 1-15 carbon atoms,  $\text{R}_{18}$  and  $\text{R}_{19}$  are each an alkyl group having 1-14 carbon atoms, e and f are each an integer of 0 or 1 and n is an integer of 0-30, and



a pentaerythritol dicarboxylic acid ester represented by the general formula



wherein  $\text{R}_{22}-\text{R}_{27}$  are each an alkyl group having 3-15 carbon atoms,  $\text{R}_{28}$  is a divalent hydrocarbon group having 1-8 carbon atoms and  $h$  is an integer of 1-5.

These conventional oils may be used singly or jointly for adding to the refrigerator oil of this invention. Further, the refrigerator oil of this invention may be incorporated with paraffinic mineral oils, naphthenic mineral oils, poly $\alpha$ -olefins, alkylbenzenes and the like, but, in this case, the resulting mixed oil will be lowered in compatibility with chlorine-free type halogenocarbons.

The amount of these conventional base oils so incorporated is not particularly limited as far as the excellent performances of the refrigerator oil of this invention are not impaired, but the pentaerythritol ester should be present in the resulting mixed oil in a ratio of usually more than 50 % by weight, preferably not less than 70 % by weight of the total amount of the mixed oil.

The refrigerator oil composition according to this invention may be incorporated further with at least one kind of a phosphorous compound selected from the group consisting

of phosphoric esters, acid phosphoric esters, amine salts of acid phosphoric esters, chlorinated phosphoric esters and phosphorous esters, to improve the oil composition in wear resistance and load resistance. These phosphorous compounds are esters of phosphoric acid or phosphorous acid and an alkanol or a polyether type alcohol, or derivatives of the esters. The phosphoric esters are exemplified by tributyl phosphate, triphenyl phosphate and tricresyl phosphate. The acid phosphoric esters are exemplified by ditetradecyl acid phosphate, dipentadecyl acid phosphate, dihexadecyl acid phosphate, diheptadecyl acid phosphate and dioctadecyl acid phosphate. The amine salts of acid phosphoric esters are exemplified by salts of the above acid phosphoric esters and amines such as methylamine, ethylamine, propylamine, butylamine, pentylamine, hexylamine, heptylamine, octylamine, dimethylamine, diethylamine, dipropylamine, dibutylamine, dipentylamine, dihexylamine, diheptylamine, dioctylamine, trimethylamine, triethylamine, tripropylamine, tributylamine, tripentylamine, trihexylamine, triheptylamine and trioctylamine. The chlorinated phosphoric esters are exemplified by tris-dichloropropyl phosphate, tris-chloroethyl phosphate, polyoxyalkylene bis[di(chloroalkyl)] phosphate and tris-chlorophenyl phosphate. The phosphorous esters are exemplified by dibutyl phosphite, tributyl phosphite, dipentyl phosphite, tripentyl phosphite, dihexyl phosphite, trihexyl phosphite, diheptyl phosphite, triheptyl phosphite,

dioctyl phosphite, trioctyl phosphite, dinonyl phosphite, didecyl phosphite, diundecyl phosphite, triundecyl phosphite, didodecyl phosphite, tridodecyl phosphite, diphenyl phosphite, triphenyl phosphite, dicresyl phosphite, tricresyl phosphite and mixtures thereof. These phosphorous compounds may be added to the refrigerator oil in a ratio of 0.1-5.0 % by weight, preferably 0.2-2.0 % by weight, of the total amount of the refrigerator oil.

To further improve the refrigerator oil of the present invention in stability, it may be incorporated with at least one kind of an epoxy compound selected from the group consisting of phenylglycidyl ether type epoxy compounds, glycidyl ester type epoxy compounds, epoxidized fatty acid monoesters and epoxidized vegetable oils. The said phenylglycidyl ether type epoxy compounds used herein include phenylglycidyl ether and alkylphenylglycidyl ethers. The said alkylphenylglycidyl ethers are those having 1 to 3 alkyl groups having 1 to 13 carbon atoms, among which are preferred those having an alkyl group having 4 to 10 carbon atoms, such as butylphenylglycidyl ether, pentylphenylglycidyl ether, hexylphenylglycidyl ether, heptylphenylglycidyl ether, octylphenylglycidyl ether, nonylphenylglycidyl ether and decylphenylglycidyl ether. The said glycidyl ester type epoxy compounds include phenylglycidyl esters, alkylglycidyl esters and alkenylglycidyl esters with glycidyl benzoate, glycidyl

acrylate, glycidyl methacrylate and the like being preferred.

The epoxidized fatty acid monoesters include esters of an epoxidized fatty acid having 12 to 20 carbon atoms and an alcohol having 1 to 8 carbon atoms, phenol or an alkylphenol. In particular, butyl, hexyl, benzyl, cyclohexyl, methoxyethyl, octyl, phenyl or butylphenyl esters of epoxidized stearic acid may preferably be used.

The epoxidized vegetable oils include epoxidized compounds of vegetable oils such as soybean oil, linseed oil and cottonseed oil.

Among these epoxy compounds, the preferable ones include phenylglycidyl ether type epoxy compounds and epoxidized fatty acid monoesters with the former being more preferable. The most preferred are phenylglycidyl ether, butylphenylglycidyl ether and mixtures thereof.

In a case where these epoxy compounds are to be incorporated in the refrigerator oil of the present invention, it is desirable that they be incorporated therein in a ratio of 0.1-5.0 % by weight, preferably 0.2-2.0 % by weight, of the total amount of the refrigerator oil.

Of course, both of the aforementioned phosphorus compounds and epoxy compounds may be used jointly.

To further enhance the refrigerator oil of this invention in performances, the refrigerator oil may be incorporated, as required, with heretofore known additives for a refrigerator oil, which include phenol-type

antioxidants such as di-tert.-butyl-p-cresol and bisphenol A; amine-type antioxidants such as phenyl- $\alpha$ -naphthylamine and N,N-di(2-naphthyl)-p-phenylenediamine; wear resistant additives such as zinc dithiophosphate; extreme pressure agents such as chlorinated paraffin and sulfur compounds; oiliness improvers such as fatty acids; antifoaming agents such as silicone-type ones; and metal inactivators such as benzotriazole. These additives may be used singly or jointly. The total amount of these additives added is ordinarily not more than 10 % by weight, preferably not more than 5 % by weight, of the total amount of the refrigerator oil.

The refrigerator oils of this invention comprising the pentaerythritol as the base oil should have such viscosity and pour point as those which are normally suitable for an ordinary refrigerator oil, but they should desirably have a pour point of not higher than  $-10^{\circ}\text{C}$ , preferably  $-20^{\circ}\text{C}$  to  $-80^{\circ}\text{C}$ , to prevent them from solidification at a low temperature. Further, they should desirably have a kinematic viscosity of not less than 2cSt, preferably not less than 3cSt at  $100^{\circ}\text{C}$ , to keep the sealability of the compressor when used, while they should desirably have a kinematic viscosity of not more than 150cSt, preferably not more than 100cSt at  $100^{\circ}\text{C}$ , in view of their fluidity at a low temperature and the efficiency of heat exchange in the evaporator when used.

The refrigerator oils of the present invention are very excellent in compatibility with chlorine-free type halogenocarbons as compared with the heretofore known refrigerator oils. The chlorine-free type halogenocarbons are exemplified by 1,1,2,2-tetrafluoroethane (HFC-134), 1,1,1,2-tetrafluoroethane (HFC-134a), 1,1-difluoroethane (HFC-152a) and trifluoromethane (HFC-23) with HFC-134a being preferred.

Further, the refrigerator oils of the present invention are excellent ones which have not only high compatibility with the chlorine-free type halogenocarbons and high electrical insulating property but also high lubricity and low hygroscopicity.

The refrigerator oils of the present invention may particularly preferably be used in refrigerators, air-conditioners, dehumidifiers, cold-storage chests, freezers, freeze and refrigeration warehouses, automatic vending machines, showcases, cooling units in chemical plants, and the like which have a reciprocating or rotary compressor. Further, the above refrigerator oils may also preferably be used in refrigerators having a centrifugal compressor.

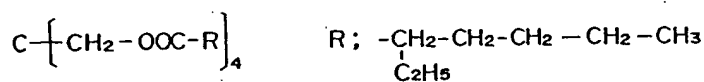
[Examples]

This invention will be better understood by the following Examples and Comparative Examples.

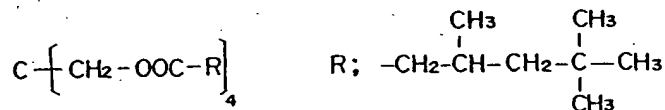
Examples 1-7 and Comparative Examples 1-6

Refrigerator oils of this invention and comparative refrigerator oils of a conventional type are illustrated as follows:

[Example 1] A tetraester of pentaerythritol (1 mol) and 2-ethylhexanoic acid (4 mol):

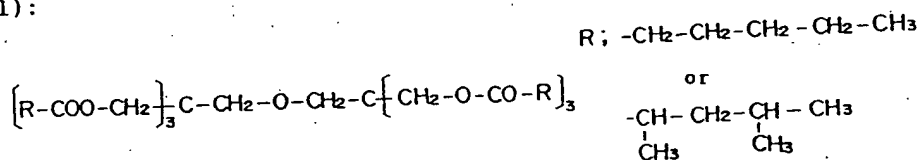


[Example 2] A tetraester of pentaerythritol (1 mol) and 3,5,5-trimethylhexanoic acid (4 mol):

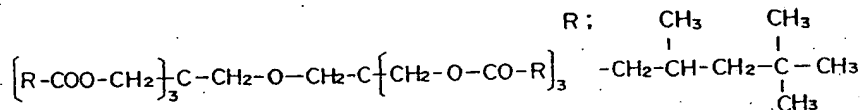


[Example 3] A tetraester of pentaerythritol (1 mol), 2-ethylhexanoic acid (2 mol) and 3,5,5-trimethylhexanoic acid (2 mol)

[Example 4] A hexaester of dipentaerythritol (1 mol), n-hexanoic acid (3 mol) and 2,4-dimethylpentanoic acid (3 mol):

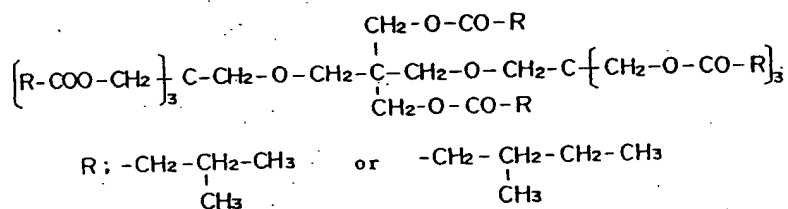


[Example 5] A hexaester of dipentaerythritol (1 mol) and 3,5,5-trimethylhexanoic acid (6 mol):



[Example 6] A mixture of 50 parts by weight of the same ester as in Example 1 and 50 parts by weight of the same ester as in Example 5.

[Example 7] A mixture of 30 parts by weight of the same ester as in Example 2, 40 parts by weight of the same ester as in Example 5 and 30 parts by weight of the following octaester of tripentaerythritol (1 mol), 3-methylbutanoic acid (4 mol) and 3-methylpentanoic acid (4 mol):



[Comparative Example 1] A naphthenic mineral oil (kinematic viscosity at 100°C: 5.2 cSt)

[Comparative Example 2] A branched-chain type alkylbenzene (kinematic viscosity at 100°C: 5.0 cSt)

[Comparative Example 3] Polyoxypropylene glycol monobutyl ether (kinematic viscosity at 100°C: 5.4 cSt)

[Comparative Example 4] Polyoxypropylene glycol dimethyl ether (kinematic viscosity at 100°C: 9.5 cSt)



[Comparative Example 5] A tetraester of pentaerythritol (1 mol) and n-nonanoic acid (4 mol)

[Comparative Example 6] A tetraester of pentaerythritol (1 mol) and coconut oil

The base oils of Examples 1-7 for the refrigerator oils of the present invention were evaluated for their performances that are their compatibility with HFC-134a, insulating property, wear resistance and hygroscopicity by the following respective test methods. For comparison, the mineral oil, the alkylbenzene, the polypropylene glycol monoalkyl ethers and the polypropylene glycol dialkyl ethers of the Comparative Examples which have heretofore been used as refrigerator oils were evaluated in the same manner as in Examples 1-7. The results are indicated in Table 1.

(Miscibility with HFC-134a)

0.2 g of the test oil of each of the Examples and the Comparative Examples and 1.8 g of the refrigerant (HFC-134a) were sealed in a glass tube having an inner diameter of 6 mm and a length of 220 mm. This glass tube was then placed in a thermostat maintained at a predetermined low temperature or high temperature to observe whether the refrigerant and the test oil were miscible with each other, separated from each other or made white-turbid.

(Insulating property)

The test oils were each measured for specific volume resistivity at 25°C in accordance with JIS C 2101.

(FALEX wear test)

The test oils were each applied to a test journal for measuring the amount of the test journal worn by having the journal run in at a test oil temperature of 100 °C under a load of 150 lb for 1 minute and then running it under a load of 250 lb for 2 hours in accordance with ASTM D 2670.

(Hygroscopicity)

Thirty grams (30 g) of each of the test oils were placed in a 300-ml beaker, allowed to stand for 7 days in an air-conditioned bath maintained at a temperature of 60 °C and a humidity of 30 % and then measured for water content by the Karl-Fischer method.

Table 1

| Example<br>Comparative<br>Example | Kinematic<br>viscosity<br>@ 100°C<br>(cSt) | Miscibility<br>with HFC-134a<br>Miscible<br>temperature<br>range (°C) | Resistivity<br>@25°C<br>(Ω·cm) | FALEX test<br>Amount of<br>journal<br>worn<br>(mg) | Hygro-<br>scopicity<br>60°C, 30%<br>(%) |
|-----------------------------------|--------------------------------------------|-----------------------------------------------------------------------|--------------------------------|----------------------------------------------------|-----------------------------------------|
| Example1                          | 6.2                                        | -27 - CT *                                                            | $4.1 \times 10^{14}$           | 27                                                 | 0.19                                    |
| Example2                          | 11.5                                       | -32 - CT                                                              | $3.8 \times 10^{14}$           | 25                                                 | 0.19                                    |
| Example3                          | 8.1                                        | -28 - CT                                                              | $4.0 \times 10^{14}$           | 26                                                 | 0.19                                    |
| Example4                          | 10.0                                       | <-70 - CT                                                             | $2.8 \times 10^{14}$           | 22                                                 | 0.17                                    |
| Example5                          | 28.2                                       | - 7 - 75                                                              | $3.0 \times 10^{14}$           | 19                                                 | 0.15                                    |
| Example6                          | 11.8                                       | -25 - 90                                                              | $3.2 \times 10^{14}$           | 22                                                 | 0.16                                    |
| Example7                          | 32.3                                       | - 5 - 86                                                              | $2.4 \times 10^{14}$           | 18                                                 | 0.13                                    |
| Com.Ex.1                          | 5.2                                        | Immiscible                                                            | -                              | -                                                  | -                                       |
| Com.Ex.2                          | 5.0                                        | Immiscible                                                            | -                              | -                                                  | -                                       |
| Com.Ex.3                          | 5.4                                        | <-70 - 96                                                             | $3.3 \times 10^{11}$           | 40                                                 | 1.31                                    |
| Com.Ex.4                          | 9.5                                        | -65 - 75                                                              | $2.8 \times 10^{11}$           | 37                                                 | 0.90                                    |
| Com.Ex.5                          | 6.1                                        | Immiscible                                                            | -                              | -                                                  | -                                       |
| Com.Ex.6                          | 4.9                                        | Immiscible                                                            | -                              | -                                                  | -                                       |

\*: CT; Critical temperature of HFC-134a (102°C)

It is apparent from the results indicated in Table i that the refrigerator oils (Examples 1-7) of the present invention are very excellent in miscibility with a refrigerant, HFC-134a, as compared with those of Comparative Examples 1-2 and 5-6.

The ester of Comparative Example 5 in which the alkyl groups of the acid moiety are all straight-chain ones is inferior in miscibility with the refrigerant. Further, the ester of Comparative Example 6 which has heretofore been used as lubricating oils, refrigerator oils and the like and which is a tetraester of pentaerythritol and a natural fat or oil, is also inferior in miscibility with the refrigerant.

The ethers of Comparative Examples 3 and 4 are excellent in miscibility with the refrigerant, but these ethers are inferior in insulating property thereby rendering them unusable for hermetic type compressors. Still further, the ethers of Comparative Examples 3-4 have hygroscopicity 5-10 times that of the refrigerator oils of Examples 1-7 and are also inferior in electrical insulating property, ice choke, wear resistance, stability and the like to the refrigerator oils of the Examples.

The FALEX wear test shows that the refrigerator oils of Examples 1-7 are at least equal in wear resistance to those of Comparative Examples 3-4.

[Effect of the Invention]

As is apparent from the above comparative experiments, lubricating oils for compressors of refrigerators of the present invention are suitable for use in refrigerators using therein a hydrogen-containing halogenocarbon as a refrigerant and are excellent in electrical insulating property which is indispensable to lubricating oils for hermetic type compressors, as well as wear resistance and nonhygroscopicity.

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